

Interactive Example Candidate Responses

Paper 42 (May/June 2016), Question 6

Cambridge International AS & A Level Chemistry 9701

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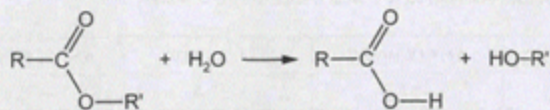
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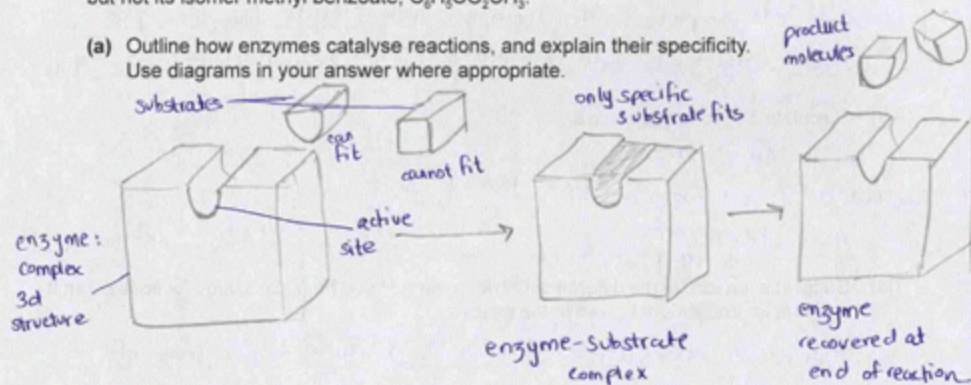
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6 Esterases are enzymes that hydrolyse esters.



Enzymes can be quite specific in the structures of the substrates they act upon. For example, an esterase isolated from the mould *Aspergillus niger* will hydrolyse phenyl ethanoate, $\text{CH}_3\text{CO}_2\text{C}_6\text{H}_5$, but not its isomer methyl benzoate, $\text{C}_6\text{H}_5\text{CO}_2\text{CH}_3$.

(a) Outline how enzymes catalyse reactions, and explain their specificity. Use diagrams in your answer where appropriate.



Enzymes are biological catalysts. They reduce the activation energy for a reaction by providing an alternative pathway for reaction. Now ~~then~~ many more substrate molecules have the E_a required and reaction rate increases. Enzymes are highly specific, with an active site only certain substrates can fit into. An enzyme-substrate complex is formed, the product is made and enzyme is recovered unchanged.

[3]

Your
Mark

6(a)

6(b)(i)

6(b)(ii)

Q6 Mark scheme

(a)	<p><i>essential mark</i></p> <p>M1: the reactants / substrate has a shape complementary / specific to active site – can be awarded from a labelled diagram as below or diagrams showing this specificity clearly <i>any two of</i></p> <p>M2: reactants / substrate binds to / fits into the <u>active site</u> of the enzyme</p> <p>M3: (Interaction with site) causes a specific bond to be weakened, (which breaks) or lowers activation energy</p> <p>M4: forms an E-S complex</p> <p>M5: products released from enzyme / active site</p> <p>labelled diagrams</p> <p>[3]</p>
(b)(i)	<p>δ 26 is CH₃-CO δ 52 is CH₃-O δ 169 is CH₃CO δ 167 is phenyl-CO</p> <p>Phenyl ethanoate is B methyl benzoate is A</p> <p>M1 = any two correct δ linked to phenylethanoate / methyl benzoate</p> <p>M2 = the rest correct</p> <p>[2]</p>
(b)(ii)	<p>heat with H_3O^+ (to hydrolyse the ester) then add $\text{Br}_2(\text{aq})$ / bromine water decolourises/gives white ppt. (with phenol from B)</p> <p>[3]</p> <p>[Total: 8]</p>

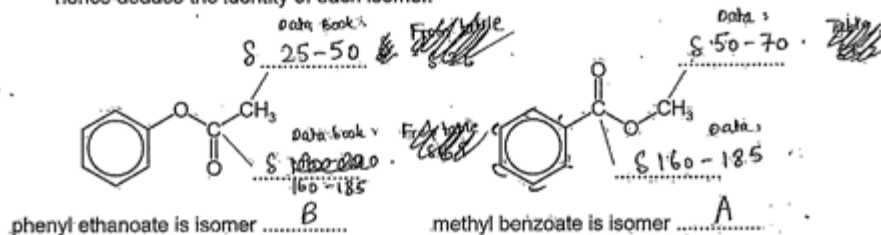
(b) Sample bottles of each of the isomers phenyl ethanoate and methyl benzoate have lost their labels and so have been named isomer A and isomer B.

(i) The carbon-13 NMR spectra of isomers A and B contain the following peaks.

isomer A	isomer B
δ 52	δ 26
δ 128	δ 122
δ 129	δ 126
δ 130	δ 129
δ 133	δ 151
δ 167	δ 169

The identity of the compound responsible for each spectrum can be deduced by studying the chemical shifts (δ) of the peaks in the spectra.

Use the Data Booklet to assign the correct peaks to the labelled carbon atoms in the structures of the isomers below. Write each value next to the relevant carbon atom and hence deduce the identity of each isomer.



[2]

(ii) These two isomers are difficult to distinguish chemically.

Describe a method of converting them to suitable products in step 1 which can then be tested in step 2.

You should state the reagents and conditions for each step, and any observations you would make.

step 1 ... hydrolyse with dil. HCl and heat. A phenol and a benzoic acid will result.

step 2 React with Br_2 (aq). The phenol will form a white ppt precipitate of 2,4,6-tribromophenol. The benzoic acid will not. So phenyl ethanoate will be the white ppt compound parent.

[3]

[Total: 8]

Your
Mark

6(a)

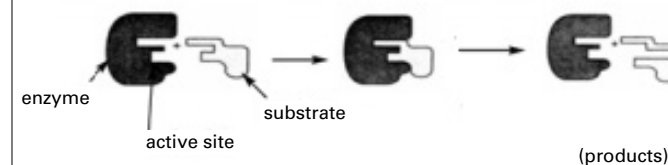
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labelled diagrams



[3]

- (b)(i) δ 26 is $\text{CH}_3\text{-CO}$ δ 52 is $\text{CH}_3\text{-O}$
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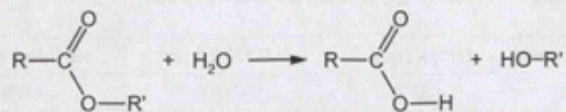
[2]

- (b)(ii) heat with H_3O^+ (to hydrolyse the ester)
 then add $\text{Br}_2(\text{aq})$ / bromine water
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[3]

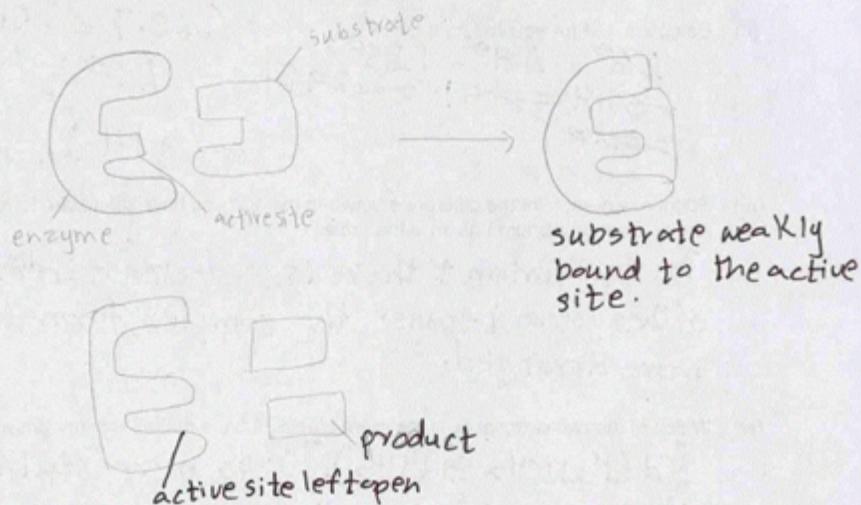
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- (a) Outline how enzymes catalyse reactions, and explain their specificity. Use diagrams in your answer where appropriate.



The enzymes have an active site with a specific three-dimensional shape which is complementary to the substrate. The substrate weakly binds to it and is converted to product after which the active site is left open for another substrate molecule. [3]

Select page

Your Mark

6(a)

6(b)(i)

6(b)(ii)

Q6 Mark scheme

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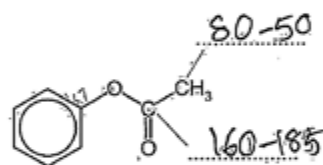
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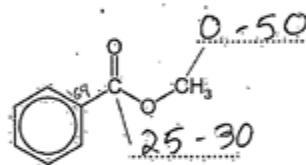
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phenyl ethanoate is isomer A



methyl benzoate is isomer B

[2]

(ii) These two isomers are difficult to distinguish chemically.

Describe a method of converting them to suitable products in step 1 which can then be tested in step 2.

You should state the reagents and conditions for each step, and any observations you would make.

step 1 Add a carboxylic acid to it. It will lead to the formation of phenol

step 2 Phenol can be tested with aq. Bromine. it will form a white ppt.

[3]

Your
Mark

6(a)

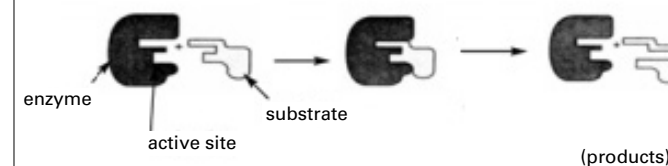
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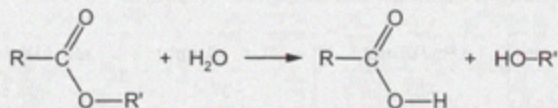
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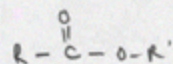
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Enzymes have a specific site called active site, which provides adsorption for specific reactant only. They act as a catalyst and speed up the reaction.

[3]

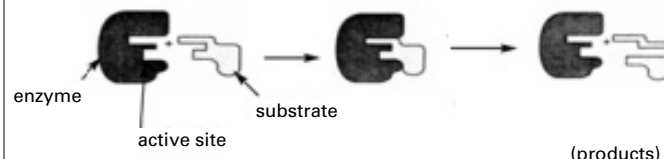
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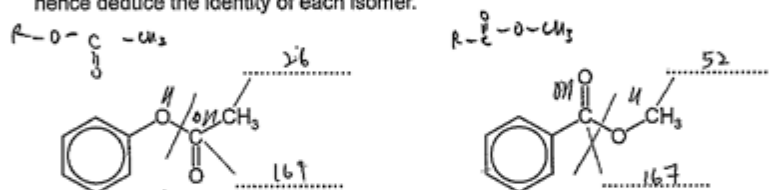
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methyl benzoate is isomer A

[2]

(ii) These two isomers are difficult to distinguish chemically.

Describe a method of converting them to suitable products in step 1 which can then be tested in step 2. You should state the reagents and conditions for each step, and any observations you would make.

step 1 ~~Use concentrated hydrochloric acid~~ Hydrolysis is used to produce 2 different products for each compound. Dilute H^+ (aq) and heat under reflux.

step 2 Use $I_2(aq) + NaOH(aq)$. The products of isomer B from step 1 will give yellow precipitate as positive result.

[3]

[Total: 8]

Your
Mark

6(a)

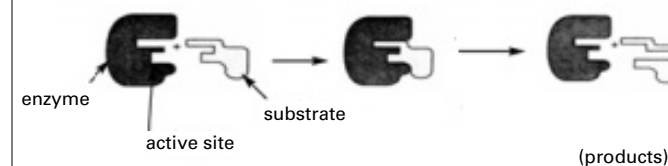
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